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Solutions to exercises 1B.1(b) The mean speed is $\sqrt{3RT/M}$ The mean translational kinetic energy is $\frac{3}{2}kT$ The ratios of species 1 to species 2 at the same temperature are $\frac{v_{\text{mean},1}}{v_{\text{mean},2}} = \sqrt{\frac{M_2}{M_1}}$ (i) $\frac{v_{\text{mean},H}}{v_{\text{mean},Hg}} = \sqrt{\frac{200.6}{4.003}} = 7.079$ (ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

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